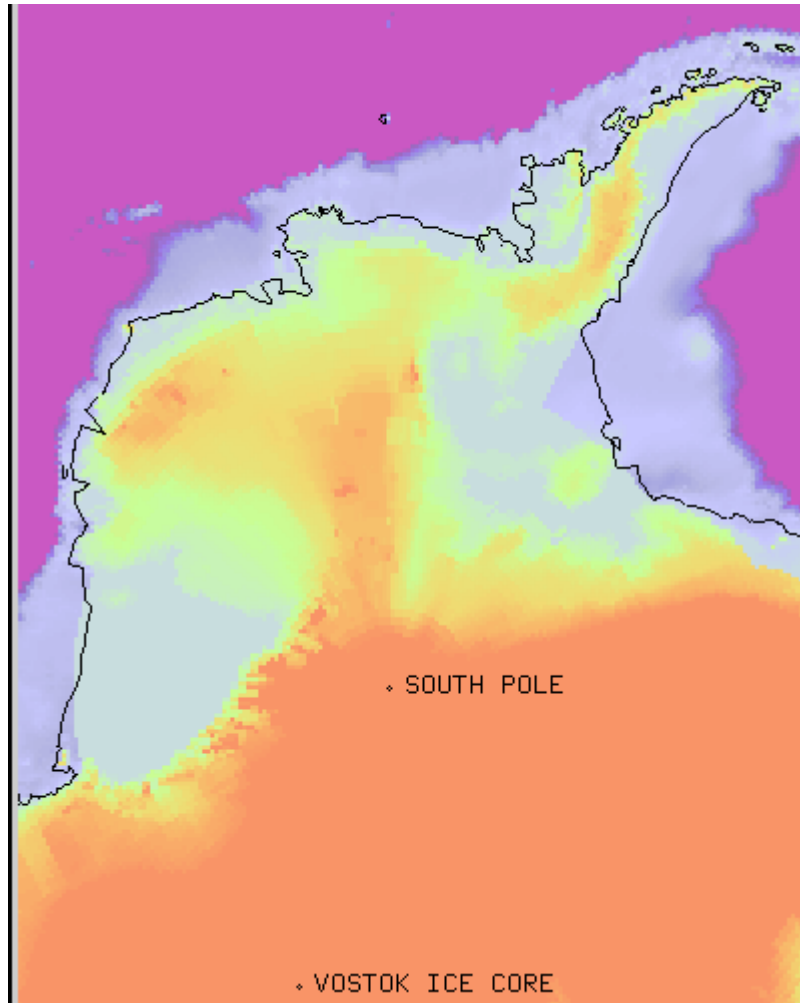


# The Climate System

## Lab: Vostok Ice Core



Map showing Antarctica and location of Vostok Ice Core. Colors indicate elevation.

### I. Introduction

Snow falling in the polar regions of the earth (e.g. Greenland and Antarctica) sometimes is preserved as annual layers within the ice sheets, provided that they are not destroyed by flow of the ice. These annual layers provide a record of the earth's climate that reaches back as much as 200,000 years.

Several different climate indicators can be measured from samples of the ice:

- The amount of dust in each annual layer is indicative of the environment at the time that the dust was deposited. Various kinds of fallout from the atmosphere, including airborne continental dust and biological material, volcanic debris, sea salts, cosmic particles, and isotopes produced by

cosmic radiation, are deposited on the ice sheet surface along with the snow, thus mixing with the snow and also acting as a distinctive barrier between different ice layers.

- The composition of bubbles of air trapped in the ice is a measure of the composition of the atmosphere in ancient times. With increasing pressure from subsequent snow deposition on an ice cap or glacier, the snow becomes compacted and, consequently, air is trapped within the deposited layer. This entrapment of air occurs essentially with no differentiation of the atmospheric gas components. However, carbon dioxide has different chemical properties from other atmospheric gases, thus, the carbon dioxide concentration in the air-filled spaces might be affected by interaction with the ice itself or with trapped impurities.
- The isotopic composition of water, and in particular the concentration of the heavy isotope of oxygen,  $^{18}\text{O}$ , relative to  $^{16}\text{O}$ , as well as  $^2\text{H}$  (deuterium, D) relative to  $^1\text{H}$ , is indicative of the temperatures of the environment. During cold periods, the concentration of less volatile  $^2\text{H}$  ( $^{18}\text{O}$ ) in the ice is lower than during warm periods. The reason for this is that at lower temperature, the moisture has been removed from the atmosphere to a larger degree resulting in an increased depletion of the heavier isotopes.

The Vostok core was drilled in East Antarctica, at the Soviet station Vostok from an altitude of 3488 m, and has a total length of 2083 m. Ice samples have been analyzed with respect to isotopic content in  $^2\text{H}$ , dust, and methane and carbon dioxide trapped in air bubbles. The profiles of  $^2\text{H}$ , methane, and carbon dioxide concentrations behave in a similar way with respect to depth in the core, showing a short interglacial stage, the Holocene, at the top, a long glacial stage below, and the last interglacial stage near the bottom of the core. The record goes back in time about 160,000 years.

## II. Lab Instructions

For the purposes of the lab, assume that the most shallow ice core measurements represent the environmental conditions in the 18th century before the Industrial Revolution.

### A. Gas Age vs. Ice Age

Age is calculated in two different ways within an ice core. The ice age is calculated from an analysis of annual layers in the top part of the core, and using an ice flow model for the bottom part. The gas age data accounts for the fact that gas is only trapped in the ice at a depth well below the surface where the pores close up.

**Task 1:** Transfer the Vostok ice core data (vos\_data.tsv) to Excel, and save it as an Excel workbook. Plot both the ice age and the gas age as a function of depth on the same graph. Answer the following questions:

1. What are the units of both age scales?
2. a) What are the depths of the shallowest and deepest data points?  
b) Does age increase or decrease down the core?  
c) Why?
3. Is the thickness of an annual layer of ice smallest at the top or bottom of the core? Why?

4. Note that if you look carefully at the plot you can see that the curve changes slope between the top and the bottom of the core. Why do you think that this happens?

### ***B. $\delta D$ (Change in Deuterium) as a proxy for temperature***

**Task 2:** Next you will calculate the temperature based on the isotopic composition of the ice. Insert a blank column into the table to the right of the delta-deuterium column ( $\delta D$ ). Isotopic ratios are used to model temperature. Calculate the temperature at Vostok based on the following formula describing the empirical relationship between temperature and deuterium concentration:

$$\text{Temperature (deg-C)} = -55.5 + (\delta D + 440) / 6$$

**\*Be sure to save your work!\***

Now plot your calculated temperature vs. ice age.

1. How reliable do you consider this paleoclimate record to be? (Hint: think about some of the uncertainties in the age models and compare the age of the Last Glacial Maximum (LGM) to what scientists consider the age to be today.) The Last Glacial Maximum occurred when the ice sheets were at their most recent maximum extension.
2. How long ago did the maximum temperature occur?
3. How long ago did the minimum temperature occur?
4. How do these temperatures compare to the current **average** Vostok temperature?

### ***C. $CO_2$ , $CH_4$ , and Dust***

**Task 3:** Plot  $CO_2$  as a function of gas age.

5. How closely does the plot of  $CO_2$  resemble that of temperature?

Now plot  $CO_2$  against temperature. Add a trendline and record the  $R^2$  value.

6. Do you think this correlation is significant?

**Task 4:** Make the same plots for  $CH_4$ .

7. Is  $CO_2$  or  $CH_4$  more closely correlated with temperature?
8. Why do you think that is?

Now make a plot of dust as a function of ice age. Compare this to the temperature plot.

9. How well do the changes in dust concentration correlate with the temperature changes?

**Task 5:** The present atmospheric  $CO_2$  and  $CH_4$  concentrations are 395 ppm and 1,827 ppb, respectively, according to CDIAC (2014). Fill in the chart by calculating the difference in  $CO_2$  and  $CH_4$  concentrations between the LGM and the 18th century, and between the 18th century and today (use current values given above).

Carbon Dioxide Concentrations (ppm)				Methane Concentrations (ppm)			
LGM	18 <sup>th</sup> C	Present	Difference	LGM	18 <sup>th</sup> C	Present	Difference
		X				X	
X				X			

10. Why were CO<sub>2</sub>, CH<sub>4</sub>, and dust concentrations different during the glacial time as compared to the 18th century?

**Task 6:** Insert today's CO<sub>2</sub> concentration (use the CDIAC value 395 ppm) into the equation from Task 3 to determine what the past relationship between CO<sub>2</sub> and temperature predicts that today's temperature should be at Vostok.

11. How does your calculation compare with the known value from Part B4?

#### ***D. Trends***

**Task 7:** Note that there were two major warming events representing two deglaciations in the Vostok calculated temperature data. Then look at how CO<sub>2</sub> and CH<sub>4</sub> change during those deglaciation periods.

12. From the data provided in this lab, can you tell which changes first, temperature or greenhouse gas (CO<sub>2</sub>, CH<sub>4</sub>) concentrations?

13. Why is this important?

### **III. Lab Report Instructions**

Email your graphs along with the answers to the thirteen questions in a word document. Make sure that the graphs generated are the same size for the horizontal axis for easier comparison. In addition, address the following questions:

1. How do the glacial/interglacial changes in temperature, carbon dioxide, and methane compare to the changes since the 18th century? See document "Changes since the 18<sup>th</sup> Century"
2. Why are these ice core paleoclimate records so important to our understanding and prediction of climate change?